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Addition/Correction

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Correction to Highly Sensitive NH_3 Detection Based on Organic Field-Effect Transistors with Tris(pentafluorophenyl)borane as Receptor

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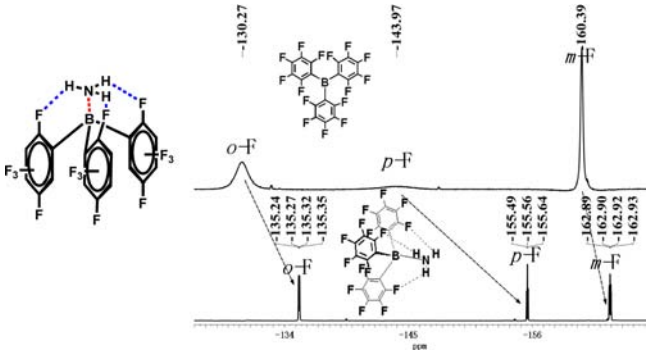
Page 14651. The chemical shift in the ^{19}F NMR spectrum of free tris(pentafluorophenyl)borane (top ^{19}F NMR spectrum in corrected Scheme 1) was influenced by the water (about 640 ppm) in ordinary C_6D_6. Therefore, we would like to offer corrected spectra, taken in anhydrous C_6D_6 (<10 ppm water).

Scheme 1. NH_3−TPFB Interaction and ^{19}F NMR Spectra of TPFB and the TPFB−NH3 Complex Synthesized and Isolated from Toluene−Chloroform Solution[a]



[a]Blue denotes hydrogen bonding, and red denotes B−N interaction.

The tris(pentafluorophenyl)borane purity is nominally 95% (from Sigma Aldrich) and was handled in a glovebag filled with dry nitrogen and stored in dynamic vacuum. Due to the possibility that the 5% impurity may be water−borane or other adduct and may be exchanging in solution, the peaks in the tris(pentafluorophenyl)borane ^{19}F NMR spectrum appear broad and slightly shifted. The ^{19}F NMR peaks for 100% pure tris(pentafluorophenyl)borane in C_6D_6 are available in the references given below: δ −129.1(*o*-F), −142.0 (*p*-F), −160.3 (*m*-F). The conclusions originally presented regarding response of devices to ammonia are unaffected by this revision; in fact, the spectra provide additional confirmation that the compound used in the published study was indeed anhydrous.

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